

# GUJARAT TECHNOLOGICAL UNIVERSITY

# Bachelor of Pharmacy Subject Code: BP403TP SEMESTER: IV

Subject Name: Physical Pharmaceutics II

**Scope**: The course deals with the various physica and physicochemical properties, and principles involved in dosage forms/formulations. Theory and practical components of the subject help the student to get a better insight into various areas of formulation research and development, and stability studies of pharmaceutical dosage forms.

Objectives: Upon completion of the course the student shall be able to

- 1. Understand various physicochemical properties of drug molecules in the designing the dosage forms
- 2. Know the principles of chemical kinetics & to use them for stability testing nad determination of expiry date of formulations
- 3. Demonstrate use of physicochemical properties in the formulation development and evaluation of dosage forms

# Teaching scheme and examination scheme:

Teaching Scheme				Evaluation Scheme			
Theory	Tutorial	Practical	Total	Theory		Practical	
				External	Internal	External	Internal
3	1	4	6	80	20	80	20

#### **Course Content:**

Sr No	Topics		
		weightage	
1.	Colloidal dispersions: Classification of dispersed systems & their general	7	
	characteristics, size & shapes of colloidal particles, classification of colloids &		
	comparative account of their general properties. Optical, kinetic & electrical		
	properties. Effect of electrolytes, coacervation, peptization& protective action		
2.	Rheology: Newtonian systems, law of flow, kinematic viscosity, effect of	10	
	temperature, non-Newtonian systems, pseudoplastic, dilatant, plastic,		
	thixotropy, thixotropy in formulation, determination of viscosity, capillary,		
	falling Sphere, rotational viscometers		
	<b>Deformation of solids:</b> Plastic and elastic deformation, Heckel equation, Stress,		
	Strain, Elastic Modulus	1.0	
3.	Coarse dispersion: Suspension, interfacial properties of suspended particles,	10	
	settling in suspensions, formulation of flocculated and deflocculated		
	suspensions. Emulsions and theories of emulsification, microemulsion and		
	multiple emulsions; Stability of emulsions,		
	preservation of emulsions, rheological properties of emulsions and emulsion		
	formulation by HLB method	10	
4	Micromeretics: Particle size and distribution, mean particle size, number and	10	
4.	weight distribution, particle number, methods for determining particle size by		
	different methods, counting and separation method, particle shape, specific		
	surface, methods for determining surface area, permeability, adsorption, derived		
	properties of powders, porosity, packing arrangement, densities, bulkiness &		
5.	flow properties. <b>Drug stability:</b> Reaction kinetics: zero, pseudo-zero, first & second order, units	10	
<i>J</i> .	of basic rate constants, determination of reaction order. Physical and chemical	10	
	factors influencing the chemical degradation of pharmaceutical product:		
	temperature, solvent, ionic strength, dielectric constant, specific & general acid		
	base catalysis, Simple numerical problems. Stabilization of medicinal agents		
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against common reactions like hydrolysis & oxidation. Accelerated stability						
testing in expiration dating of pharmaceutical dosage forms. Photolytic						
degradation and its prevention						

# PHYSICAL PHARMACEUTICS- II (Practical)

- 1. Determination of particle size, particle size distribution using sieving method
- 2. Determination of particle size, particle size distribution using Microscopic method
- 3. Determination of bulk density, true density and porosity
- 4. Determine the angle of repose and influence of lubricant on angle of repose
- 5. Determination of viscosity of liquid using Ostwald's viscometer
- 6. Determination sedimentation volume with effect of different suspending agent
- 7. Determination sedimentation volume with effect of different concentration of
- 1. single suspending agent
- 8. Determination of viscosity of semisolid by using Brookfield viscometer
- 9. Determination of reaction rate constant first order.
- 10. Determination of reaction rate constant second order
- 11. Accelerated stability studies

#### **Recommended Books: (Latest Editions)**

- 1. Physical Pharmacy by Alfred Martin, Sixth edition
- 2. Experimental pharmaceutics by Eugene, Parott.
- 3. Tutorial pharmacy by Cooper and Gunn.
- 4. Stocklosam J. Pharmaceutical calculations, Lea & Febiger, Philadelphia.
- 5. Liberman H.A, Lachman C., Pharmaceutical Dosage forms, Tablets, Volume-1 to 3, Marcel Dekkar Inc.
- 6. Liberman H.A, Lachman C, Pharmaceutical dosage forms. Disperse systems, volume 1, 2, 3. Marcel Dekkar Inc.
- 7. Physical Pharmaceutics by Ramasamy C, and Manavalan R.