GUJARAT TECHNOLOGICAL UNIVERSITY

B.Pharm SEMESTER: I

Subject Name: PHARMACEUTICAL INORGANIC CHEMISTRY

Subject Code: BP104TP

Scope: This subject deals with the monographs of inorganic drugs and pharmaceuticals

Objectives: Upon completion of course student shall be able to

1. know the sources of impurities and methods to determine the impurities in inorganic drugs and pharmaceuticals

2. understand the medicinal and pharmaceutical importance of inorganic compounds

Sr No	Course Contents	Total Hrs
1	Impurities in pharmaceutical substances: History of Pharmacopoeia,	10
	Sources and types of impurities, principle involved in the limit test for	
	Chloride, Sulphate, Iron, Arsenic, Lead and Heavy metals, modified limit test	
	for Chloride and Sulphate	
	General methods of preparation, assay for the compounds superscripted with	
	asterisk (*), properties and medicinal uses of inorganic compounds belonging	
	to the following classes	
2	Acids, Bases and Buffers: Buffer equations and buffer capacity in general,	10
	buffers in pharmaceutical systems, preparation, stability, buffered isotonic	
	solutions, measurements of tonicity, calculations and methods of adjusting	
	isotonicity.	
	Major extra and intracellular electrolytes: Functions of major physiological	
	ions, Electrolytes used in the replacement therapy: Sodium chloride*,	
	Potassium chloride, Calcium gluconate* and Oral Rehydration Salt (ORS),	
	Physiological acid base balance.	
	Dental products : Dentifrices, role of fluoride in the treatment of dental caries,	
	Desensitizing agents, Calcium carbonate, Sodium fluoride, and Zinc eugenol	
3	cement.	10
3	Gastrointestinal agents Acidifiers: Ammonium chloride* and Dil. HCl	10
	Antacid: Ideal properties of antacids, combinations of antacids, Sodium	
	Bicarbonate*, Aluminum hydroxide gel, Magnesium hydroxide mixture	
	Cathartics: Magnesium sulphate, Sodium orthophosphate, Kaolin and	
	Bentonite	
	Antimicrobials: Mechanism, classification, Potassium permanganate, Boric	
	acid, Hydrogen peroxide*, Chlorinated lime*, Iodine and its preparations	
4	Miscellaneous compounds	8
•	Expectorants: Potassium iodide, Ammonium chloride*.	
	Emetics: Copper sulphate*, Sodium potassium tartarate	
	Haematinics: Ferrous sulphate*, Ferrous gluconate	
	Poison and Antidote: Sodium thiosulphate*, Activated charcoal, Sodium	
	nitrite333	
	Astringents: Zinc Sulphate, Potash Alum	
5	Radiopharmaceuticals: Radio activity, Measurement of radioactivity,	7
	Properties of α , β , γ radiations, Half life, radio isotopes and study of radio	
	isotopes - Sodium iodide I131, Storage conditions, precautions &	
	pharmaceutical application of radioactive substances.	

Practical

I Limit tests for following ions

Limit test for Chlorides and Sulphates

Modified limit test for Chlorides and Sulphates

Limit test for Iron

Limit test for Heavy metals

Limit test for Lead

Limit test for Arsenic

II **Identification test** Magnesium hydroxide Ferrous sulphate Sodium bicarbonate Calcium gluconate Copper sulphate

III Test for purity

Swelling power of Bentonite

Neutralizing capacity of aluminum hydroxide gel

Determination of potassium iodate and iodine in potassium Iodide

IV Preparation of inorganic pharmaceuticals

Boric acid Potash alum Ferrous sulphate

Recommended Books (Latest Editions)

- 1. A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4th edition.
- 2. A.I. Vogel, Text Book of Quantitative Inorganic analysis
- 3. P. Gundu Rao, Inorganic Pharmaceutical Chemistry, 3rd Edition
- 4. M.L Schroff, Inorganic Pharmaceutical Chemistry
- 5. Bentley and Driver's Textbook of Pharmaceutical Chemistry
- 6. Anand & Chatwal, Inorganic Pharmaceutical Chemistry
- 7. Indian Pharmacopoeia